THE PHOTOCHEMISTRY OF A PHENYLPROPENYL ALLENE; SUPPRESSION OF THE DI- π -METHANE REARRANGEMENT IN AN ALLENIC SYSTEM

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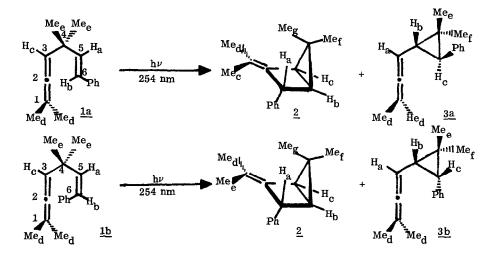
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Recently, several studies have appeared² which focus on the question of the importance of C-4 substitution on the singlet di- π -methane rearrangement. It has been demonstrated that the presence of <u>gem</u>-dimethyl substitution at C-4 appears to favor the di- π -methane rearrangement² (vinyl cyclopropane formation) while the absence of methyls and the presence of hydrogens at C-4 tend to favor formation of bicyclo[2.1.0]-pentane (housane) derivatives.^{2a, b, c} As a continuation of our studies involving the di- π -methane rearrangement³, we have had occasion to study the effect of unusual hybridization at C-2 on the course of the rearrangement. We wish to report our preliminary results on the photochemistry of the allenes <u>1a</u> and <u>1b</u> where it is observed that the di- π -methane rearrangement, specifically of <u>1a</u>, is suppressed in favor of what is presumed to be the symmetry allowed $\frac{2}{\pi s} + \frac{2}{\pi s}$ intramolecular cycloaddition.

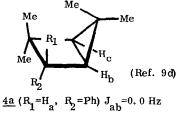


Direct irradiation⁴ of $\underline{1a}^{5, 6a, b}$ ($J_{ab} = 15.5 \text{ Hz}$)^{6C} in <u>t</u>-butyl alcohol for 5 min (7.5% conversion) gives primarily the bicyclo[2.10] pentane $\underline{2}^{6}$ and a minor product assigned the allene cyclopropane structure <u>3a</u> (4.9:1, respectively). Similarly, direct irradiation of <u>1b</u> ($J_{ab} = 12.5 \text{ Hz}$) gives predominantly <u>2</u> along with a compound tentatively assigned <u>3b</u>^{7a}, the epimer of <u>3a</u>^{7b}. For preparative purposes irradiations were conducted in benzene (30 min) where similar results were obtained although the ratio <u>2:3a</u> is reduced to 1.9:1. In contrast to the direct irradiation studies, sensitized irradiation using acetone as a solvent sensitizer gave rise to geometric isomerization (cis:trans from <u>1a</u>, 7.5:1; from <u>1b</u>, 5.6:1) without skeletal reorganization. This indicates that both the di- π -methane rearrangement products <u>3a</u> and <u>3b</u> and the bicyclo[2.1.0]pentane <u>2</u> are formed by way of the singlet rather than triplet state. Furthermore all appear to be primary photoproducts based upon data obtained in the low conversion runs.

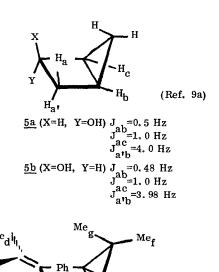
The structures of the photoproducts were established using pmr and ir spectral data. Their mass spectra confirm that they are isomeric with the precursor allenes. Retention of the allene moiety and loss of the styryl chromophore in <u>3a</u> and <u>3b</u> were confirmed by ir and pmr spectroscopy. The long range coupling between Me_d and H_a (J_{ad} =3.0 Hz), observed in the pmr spectra of <u>3a</u> and <u>3b</u>, as well as <u>1a</u> and <u>1b</u> (J_{cd} =3.0 Hz), is typical for related allenic systems. ^{5b} The stereochemistry tentatively assigned to <u>3a</u> was established on the basis of the pmr double resonance experiments (J_b =6.5 Ha)⁸ and nuclear Overhauser effect studies.

The structure $\underline{2}$ was arrived at by examination of the pmr spectrum which, aside from the aromatic protons, showed a broadened singlet at $\delta 3.17$ (H_a), a broadened doublet at 2.15 (H_c), a sharp doublet centered at 1.61 (H_b, $J_{ab} < 0.5$ Hz; $J_{bc} = 5.0$ Hz), in addition to four singlets at $\delta 1.60$, 1.24, 1.21 and 1.01 assigned to Me_d , Me_b , Me_f and Me_g , respectively. The two lower field singlets are broadened and of lesser intensity relative to the two higher field singlets and this may be attributed to long range homoallylic coupling to H_a. This observation was confirmed by double resonance experiments. Also, the protons whose signals are assigned to H_a and H_c were shown by double resonance to be coupled. The anti configuration assigned to $\underline{2}$ is consistent with the reported data since coupling has been observed for the analogous protons in closely related systems such as $\underline{4} - \underline{6}$. ⁹ Additional confirmation of this structural assignment was obtained utilizing the nuclear Overhauser effect. NOe's of 13.3% and 4.5% were observed between the signals assigned to Me_g and H_a , and Me_e and H_a , respectively. Furthermore, if the bicyclo[2.1.0]pentane possessed the alternate epimeric structure T_a a coupling constant (J_{ab}) of 2 - 4 Hz would be expected as is observed in the structurally similar model compounds $\underline{4} - \underline{6}$. The thermal reversion of $\underline{2}$ to $\underline{1a}$ (T>200^oC), a reaction which is not without precedent in bicyclo[2.1.0]pentane chemistry. ¹⁰ provides additional support for the assignment.

Several points regarding the preliminary observations presented here deserve comment. The unusual feature of the photochemistry observed for <u>1a</u> and <u>1b</u> is that the cycloaddition reaction leading to <u>2</u>, represents the major reaction pathway and that the di- π -methane process, affording <u>3a</u> and <u>3b</u>, represents the minor reaction mode. These results are to be contrasted with those obtained for the structurally related acyclic 1, 4-diolefinic systems^{2a}, where vinyl cyclopropane rather than bicyclo[2.1.0] pentane formation is the

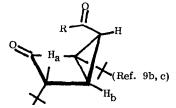


 $\underline{4b}$ (R₁=Ph, R₂=H_a) J_{ab}=2.0 Hz



н_ь

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 $\underline{6a}$ (R=Ph) J_{ab} =0.0 Hz <u>6b</u> (R=<u>t</u>-Bu) J_{ab} =0.0 Hz

major reaction pathway.¹¹ Furthermore, it has been asserted^{2c} that in the absence of gem-dimethyl substitution at the C-atom flanked by the π -moieties, bicyclo[2.1.0] pentane formation should dominate in 1, 4-pentadienyl systems. In the case of <u>1a</u>, the fact that only <u>3a</u> is formed at low conversion parallels the previous observations in related systems $^{2a, c, 3a, 12}$ that di- π -methane rearrangement proceeds in a highly stereoselective manner and suggests the reaction may be concerted. In addition, the formation of the bicyclo[2.1.0] pentane 2 having the anti configuration from 1a and 1b is consistent with principles based upon orbital symmetry considerations.¹³ Two possible modes of disrotatory cyclization are available for <u>1a</u> and <u>1b</u> and stereochemical arguments may be invoked to account for thermodynamic control leading to preferential formation of 2; however, cyclization by sequential bond formation involving diradical intermediates is also mechanistically plausible. Finally, the formation of $\underline{2}$, having the anti configuration, as the sole bicyclic photoproduct from 1a and 1b, is to be contrasted with the formation of only 4b having the syn configuration from the related 1, 4-diolefinic system.^{2a} Possibly this apparent inconsistency may reflect subtle electronic and/or steric differences between the olefinic^{2a} and allenic moieties.

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- (4) Irradiations were conducted in serum-capped Vycor test tubes employing a Rayonet RPR-100 Chamber Reactor equipped with 16 8-watt 254 nm or 350 nm lamps and a "merry-go-round" apparatus (The Southern New England Ultraviolet Co., Middletown, Conn.). All samples were degassed by purging with nitrogen for a period of 45 min prior to irradiation.
- (5) (a) Compounds <u>la</u> and <u>lb</u> were obtained as a~1:1 mixture from 2, 2, 5-trimethyl-3, 4-hexadien-1-al^{5b} using the Wittig reaction with benzylidenetriphenylphosphorane. The analytical and spectral properties of <u>la</u> and <u>lb</u> are consistent with their assigned structures. Complete synthetic details will be presented in our full paper; (b) R. S. Bly and S. U. Koock, <u>J. Amer. Chem. Soc.</u>, <u>91</u>, 3292 (1969).
- (6) (a) Satisfactory combustion analyses and mass spectra were obtained for all new compounds with the exception of <u>3b</u>; (b) all separations and purifications were achieved by glc utilizing either a 5 ft or 10 ft column of 30% SE 30 on Chromasorb P at a temperature of 140[°]; (c) all pmr data were obtained at both 60 and 100 MHz.
- (7) (a) The photochemistry of <u>1b</u> is considerably more complex than the photochemistry of <u>1a</u>. This is presumably due to the prior cis-trans isomerization of <u>1b</u> to <u>1a</u> which competes effectively with the rearrangement; however, during low conversion runs using <u>1b</u>, formation of <u>2</u> appears to predominate over the di-π-methane process; (b) while compound <u>3b</u> has not been obtained pure, its nmr spectrum, as expected, was very similar to <u>3a</u>, with distinct differences appearing in the aliphatic region.
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